

**MARK SCHEME for the October/November 2011 question paper
for the guidance of teachers**

0620 CHEMISTRY

0620/52

Paper 5 (Practical), maximum raw mark 40

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

- Cambridge will not enter into discussions or correspondence in connection with these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2011 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

Page 2	Mark Scheme: Teachers' version	Syllabus	Paper
	IGCSE – October/November 2011	0620	52

- 1 (a) table of results for *experiment 1*
all boxes completed correctly (1)
readings to 1 d.p. (1)
comparable to supervisors (1) $\pm 3 \text{ cm}^3$ [3]
- (b) table of results for *experiment 2*
all boxes completed correctly (1)
readings to 1 d.p. (1)
comparable to supervisors (1) $\pm 3 \text{ cm}^3$ [3]
- (c) red / brown (1) [1]
- (d) blue / black (1) **not:** purple to colourless (1) **not:** clear [2]
- (e) as an indicator / to test for iodine / owtte (1) [1]
- (f) (i) experiment 2 (1) [1]
(ii) experiment 2 is 2× volume of experiment 1 (1) **allow:** almost double [1]
(iii) solution F more concentrated / stronger (1) or converse
twice / double (1) [2]
- (g) half value from table result for experiment 1 (1)
half volume of potassium iodate/iodine (1) [2]
- (h) (i) two sources of error
e.g. using a measuring cylinder to measure iodate/acid
experiment only carried out once
going past end point / owtte
not: references to human error / temperature / rinsing [2]
(ii) two meaningful improvements related to above
e.g. use a pipette / burette
repeat experiment [2]
- [Total: 20]**
- 2 (a) (i) liquid H blue and no smell (1) any reference to smell = 0
pH < 7 (1) [2]
(ii) liquid J colourless (1) **not:** clear
sweet smell / petrol / acetone / owtte (1) **not:** pungent / strong
two layers / immiscible (1) with indicator paper pH 7 or green (1) [4]

Page 3	Mark Scheme: Teachers' version	Syllabus	Paper
	IGCSE – October/November 2011	0620	52

- (b) white precipitate (1) [1]
- (c) (i) blue (1) precipitate (1)
on heating black (1) [3]
- (ii) blue precipitate (1) dissolves / solution (1)
deep / royal blue (1) [3]
- (d) (i) extinguished / goes out / no reaction (1)
not: no change / nothing happens [1]
- (ii) flame / catches fire (1) orange / yellow (1) smokey / sooty (1) max 2
orange / yellow or smokey / sooty flame = (2)
reference to glowing splint = 0 [2]
- (e) copper (1) sulfate (1)
acidic (1) water present / aqueous (1) max 2 [2]
- (f) organic (1) hydrocarbon / non polar (1)
allow: flammable / fuel / alkane **not:** alcohol [2]

[Total: 20]